

Abstract: cause toxicity in human body. mercury

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Abstract: Mercury is a non-essential mineral for human body.

Its small amounts can cause toxicity in human body. Mercury intake more than 1g is injurious to health. Studies show that oral LD₅₀ is approx.

100g for a 70kg man. More than recommended amounts can cause several problems in human body like cardiovascular diseases, Central nervous system impairment, lungs diseases and various metabolic issues. Mercury toxins occur in three forms; Organic mercury, Inorganic mercury and metallic or elemental mercury.

Organic mercury is the form of mercury which is soluble in organic solvents and it can be absorbed up to 95% in human body, whereas, Inorganic mercury is a water soluble form and it is accumulated in and released from kidney through urination. Metallic mercury is lipid soluble form of mercury and it can reach brain crossing the blood brain barrier. It also occurs in the form of vapors and can be inhaled. Major sources of mercury are dental amalgam for elemental mercury, fish for organic mercury and many preservatives for inorganic mercury. Mercuric toxicity can be treated with selenium and other antioxidants rich foods and intake of sulfur containing amino acids (cysteine and methionine). Besides these nutrition therapies some drugs are also created to treat toxicity of mercury.

Introduction: Toxicity is defined as the quality of being toxic or poisonous. In terms of nutrition it is excess of any nutrient or mineral.

Where, Toxicology is defined as a branch of science concerned with nature, effects and detection of poisons. In toxicology the physical and chemical properties of toxins are studied along with the state it results in and

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treatment. Mercury is a non-essential and toxic mineral for human body. It is ranked 3rd in the list of most toxic substances by United States Government Agency for Toxic Substances and Disease Registry (Clifton, 2007; Kevin et al., 2014). Mercury is considered to be major environmental pollutant and is widely used in agriculture, medicines and industries. Mercury is not destroyed and it circulates in ecosystems (Jung & Zheng, 2012).

Mercury is present in different sources in three forms i. e. Organic, Inorganic and metallic mercury. Toxicology Of Mercury Organic mercury is present in the forms of Ethyl Mercury and Methyl Mercury. Inorganic form of mercury occurs as mercuric (Hg^{++}) and mercurous (Hg^{+}) salts and other compounds with sulfur chlorine or oxygen in solid state.

Metallic mercury also known as elemental mercury is liquid at room temperature and vaporizes easily (Jung & Zheng, 2012; Kevin Et Al., 2012). Mercury can enter human body via ingestion, inhalation or absorbed through skin. Its intake can cause toxicity and results depend on the form of mercury taken in.

Organic forms of mercury (Me-Hg and Eth-Hg) interfere with replication, translation and transcription, haeme synthesis and cause CNS disorders. Inorganic forms due to their water soluble nature cause renal problems and also chest pain. Elemental mercury is mainly responsible for major CNS disorders like Alzheimer's disease, skin infections and lung diseases (Kevin Et Al.

, 2012; Jung & Zheng, 2012; Syverson & kaul 2014). The toxicity is treated by both MNT and drugs. States of Mercury: Mercury is present in the
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environment in three forms¹. Organic Mercury: Organic mercury includes compounds of mercury with organic functional groups like methyl (CH₃), ethyl (C₂H₅) or phenyl (C₆H₅). Phenyl Mercury is present in different medicines as preservative (Syverson & Kaur, 2014). Ethyl Mercury is in the form of thimerosal present in vaccines, and is the recent most concerned form of mercury (Clarkson & Magos, 2008; Guzzi & La Porta, 2008; Kevin Et. Al.

2014). Methyl Mercury is, due to its lipid soluble nature, easily taken up by lower organisms. It is therefore, present in edible tissues of such animals (Clarkson & Magos, 2008; Kevin Et. Al., 2014). Half-life of methyl mercury is approx. 39-70 days. (Kevin et.

Al., 2014). 2. Inorganic Mercury: Inorganic form of mercury occurs as Mercuric (Hg⁺⁺) Salts, Mercurous (Hg⁺) Salts and Mercuric chloride (calomel). Calomel is quite insoluble in water and is considered less harmful (Syverson & Kaur, 2014). Mercuric and Mercurous Salts are more water soluble and more toxic. These are easily absorbed in Gastrointestinal Tract System (Byrns & Penning, 2010; Kevin Et.

Al., 2014). Half-life of Inorganic Mercury is around 40 days (Kevin Et.

Al., 2014) 3. Metallic Mercury: Metallic mercury is commonly known as Elemental Mercury. Elemental Mercury has a low vapor pressure (2 μm Hg) but also low latent heat of Toxicology Of Mercury evaporation (295 kJ/kg). Due to this property it is vaporized easily (Byrns & Penning, 2010; Kevin Et. Al.

, 2014; Jung & Zheng, 2012). Mercury vapor is a monoatomic gas (Clarkson, 1998). It is transported to brain either by dissolving in serum or adhering to cell membranes (Bernhoft, 2011). Exposures to elemental or metallic mercury occur at workplaces or at homes (Syverson & Kaur, 2014).