Corrigendum: evaluation of general synthesis procedures for bioflavonoid—metal co...

Health & Medicine



A Corrigendum on

Evaluation of General Synthesis Procedures for Bioflavonoid-Metal

Complexes in Air-Saturated Alkaline Solutions

by Yao, Y., Zhang, M., He, L., Wang, Y., and Chen, S. (2020). Front. Chem. 8: 589. doi: 10. 3389/fchem. 2020. 00589

In the original article, there was a mistake in the legends for Figures 3, 4, 7, and 8 as published. In these Figures, marks such as a, b, c, d, e, f, and g have not been explained. The updated legends appear below.

Figure 3. (A)UV-vis spectra and(B)main peak positions of DHM refluxed for different times. (a) DHM; (b-g) DHM refluxing for 30, 60, 90, 120, 150, and 180 min.

Figure 4. (A)UV-vis spectra and(B)main peak positions of DHM stirred for different times at pH = $8.\ 2.$ (a) DHM; (b-g) DHM stirring for $1,\ 3,\ 5,\ 10,\ 20,$ and $30\ min\ in\ pH\ 8.\ 2.$

Figure 7. EPR spectra of DHM in air-, nitrogen-, and oxygen-saturated alkaline solutions. (a) DHM only; (b) DHM + DMPO; (c-e) DHM + DMPO, in nitrogen-, air-, oxygen-saturated alkaline solution.

Figure 8. EPR spectra of superoxide-anion radical generation from DMSO reacted with a base in the presence of oxygen. (a) DMSO + DMPO + O₂; (b) Na $^+$ PhO $^-$ + DMPO + O₂; (c) Na $^+$ PhO $^-$ + O₂; (d) DMSO + Na $^+$ PhO $^-$ + O₂; (e) DMSO + Na $^+$ PhO $^-$ + DMPO + O₂.

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The authors apologize for this error and state that this does not change the scientific conclusions of the article in any way. The original article has been updated.