# Chlorhexidine digluconate c34h54cl2n10o14 structure



### Contents

| Bio Activity:         |                  |  |
|-----------------------|------------------|--|
| Molecular             | C 34 H 54 Cl 2 N |  |
| Formula               | 10 O 14          |  |
| Average mass          | 897. 757 Da      |  |
| Density               |                  |  |
| Boiling Point         |                  |  |
| Flash Point           |                  |  |
| Molar<br>Refractivity |                  |  |
| Polarizability        |                  |  |
| Surface               |                  |  |
| Tension               |                  |  |
| Molar Volume          |                  |  |
| • Experim             | ental data       |  |

- Experimental data
- Predicted ACD/Labs
- Predicted ChemAxon
- Experimental Physico-chemical Properties

# • Experimental Gravity:

1.06 g/mLAlfa

Aesar41385

# • Experimental Solubility:

10 mM in

**DMSOMedChem** 

ExpressHY-B0608

DMSO:

38mg/mLMedChem

ExpressHY-B0608

### Miscellaneous

# • Safety:

**WARNING: Irritates** 

skin and eyes, not

for human

consumption Alfa

Aesar41385

# • Bio Activity:

AntibacterialMedCh

em ExpressHY-

B0608

Anti-

infectionMedChem

ExpressHY-B0608

Anti-infection;

MedChem

ExpressHY-B0608

Chlorhexidine is an

antiseptic effective

against a wide

variety of gram-

negative and gram-

positive organisms.

MedChem Express

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positive organisms.;

Target:

Antibacterial;

Chlorhexidine is a

chemical antiseptic.

It is effective on

https://assignbuster.com/chlorhexidine-digluconate-c34h54cl2n10o14-structure/

both Gram-positive

and Gram-negative

bacteria, although

it is less effective

with some Gram-

negative bacteria. It

has both

bactericidal and

bacteriostatic

mechanisms of

action, the

mechanism of

action being

membrane

disruption, not

ATPase inactivation

as previously

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useful against fungi

and enveloped

viruses, though this

has not been

extensively

investigated.

MedChem

## ExpressHY-B0608

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AntibacterialChlorh

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extensively

investigated.

Chlorhexidine is

harmful in high

concentrations, but

is used safely in low

concentrations in

many products,

such as mouthwash

and contact lens

solutions [1, 2].

MedChem

ExpressHY-B0608

Predicted data is generated using the ACD/Labs Percepta Platform –

PhysChem Module

No predicted properties have been calculated for this compound.

https://assignbuster.com/chlorhexidine-digluconate-c34h54cl2n10o14-structure/

| Density:                  |
|---------------------------|
| Boiling Point:            |
| Vapour Pressure:          |
| Enthalpy of Vaporization: |
| Flash Point:              |
| Index of Refraction:      |
| Molar Refractivity:       |
| #H bond acceptors:        |
| #H bond donors:           |
| #Freely Rotating Bonds:   |
| #Rule of 5 Violations:    |
| ACD/LogP:                 |
| ACD/LogD (pH 5. 5):       |
| ACD/BCF (pH 5. 5):        |
| ACD/KOC (pH 5. 5):        |

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|--------------------------------------------------------|--------|
| ACD/LogD (pH 7. 4):                                    |        |
| ACD/BCF (pH 7. 4):                                     |        |
| ACD/KOC (pH 7. 4):                                     |        |

Polarizability:

Polar Surface Area:

Surface Tension:

Molar Volume:

Click to predict properties on the Chemicalize site